

N*-[(*E*)-(2,4-Dinitrophenyl)methylene]aniline*Maciej Barys* and
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barys@wcheto.chem.uni.wroc.pl**Key indicators**Single-crystal X-ray study
T = 100 K
Mean $\sigma(\text{C}-\text{C})$ = 0.002 Å
R factor = 0.043
wR factor = 0.086
Data-to-parameter ratio = 13.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_{13}\text{H}_9\text{N}_3\text{O}_4$, was obtained from the condensation reaction of 2,4-dinitrobenzaldehyde and aniline. It crystallizes with two independent molecules in the asymmetric unit. In the aniline moiety, the $\text{C}=\text{N}$ and $\text{C}-\text{N}$ bond lengths are normal. The crystal packing reveals short contacts [3.374 (2)–3.632 (3) Å] between the benzene rings of neighbouring molecules, suggesting the existence of $\pi-\pi$ stacking effects.

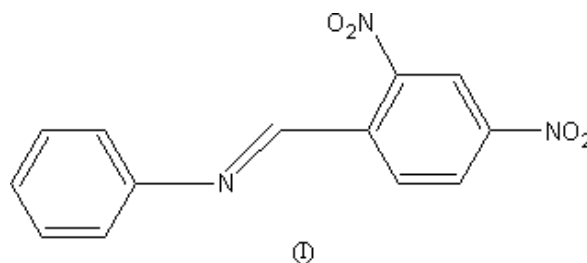
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Comment

Benzylideneanilines constitute an important class of Schiff bases that have been widely used in coordination, medical and biological chemistry over the last 25 years (Metzler *et al.*, 1980; Tarafder *et al.*, 2002). They possess significant anticancer and anti-inflammatory activity, and may also serve as reagents for stereoselective organic synthesis. Recently, the thermochromism (Pistolis *et al.*, 1996), photochromism (Jalali-Heravi, 2000) and non-linear optical properties of these compounds have found applications in modern technologies (Sekikawa *et al.*, 1997). In the design of solid materials, one of the key steps is to understand how the constituent molecules are packed, what kinds of interactions play a role in crystal packing and how they interplay with each other (Desiraju, 1989). To make a contribution in this direction, we present a crystal structure of the title compound, (I), obtained as a reaction product of aniline and 2,4-dinitrobenzaldehyde in hot methanol.



The asymmetric unit of (I) contains two independent molecules (Fig. 1), having the same bond lengths and angles within experimental error (Table 1). The average $\text{C14}=\text{N14}$ bond length of 1.270 (2) Å is close to the sum of the covalent radii of doubly-bonded C and N atoms (Holleman & Wiberg, 1995). This $\text{C}=\text{N}$ bond length is longer than that in free benzylideneaniline [1.237 (3) Å; Bürgi & Dunitz, 1970] and shorter than that found in coordinated benzylideneaniline [1.285 (10) Å; Little & Doedens, 1973]. The average (for two molecules) $\text{N14}-\text{C6}$ bond distance of 1.424 (2) Å is marginally shorter than the corresponding bond length of 1.460 (3) Å in free benzylideneaniline and close to that in the coordinated ligand of 1.428 (10) Å. The torsion angles $\text{C6A}-\text{C14A}-$

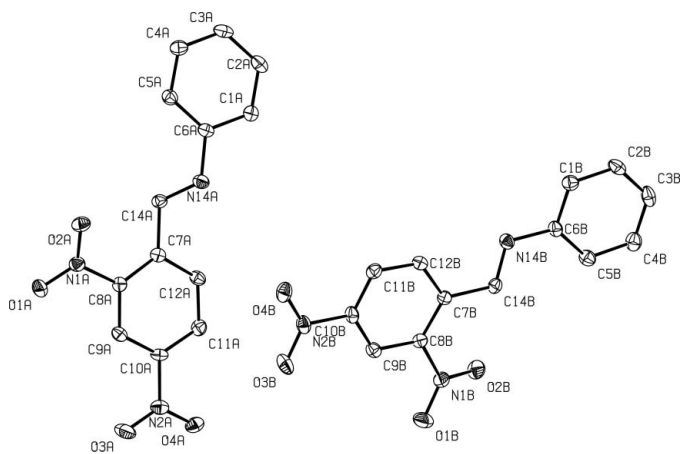


Figure 1
The two independent molecules of (I), shown with 50% probability displacement ellipsoids. H atoms have been omitted for clarity.

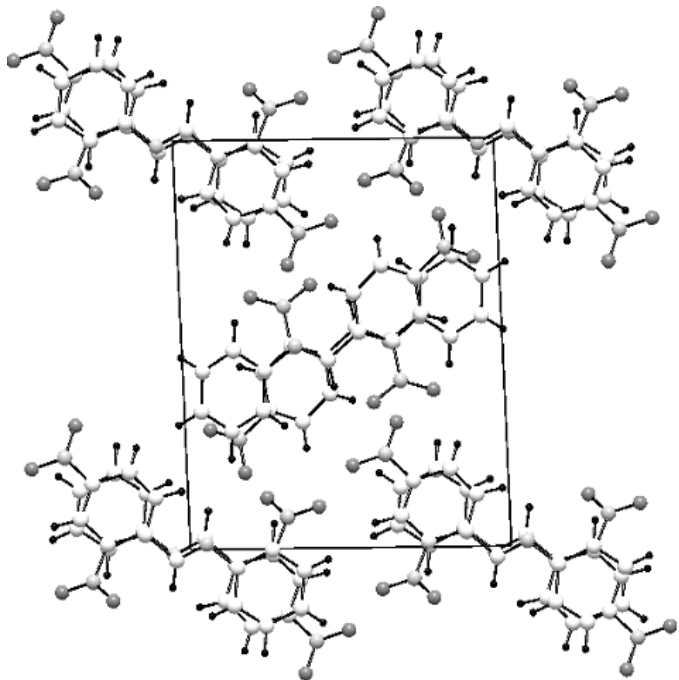


Figure 2
A packing diagram, viewed along the *a* axis.

$\text{N14A}-\text{C7A}$ [$175.30(13)^\circ$] and $\text{C6B}-\text{C14B}-\text{N14B}-\text{C7B}$ [$176.96(13)^\circ$] differ slightly for the two crystallographically independent molecules.

In the crystal structure, a partially overlapped arrangement of nearly parallel aromatic rings is observed for neighbouring independent molecules (Fig. 2). For molecule *A*, the dihedral angle and the centroid-to-centroid distance between the $\text{C1A}-\text{C6A}$ and $\text{C7A}^i-\text{C12A}^i$ benzene rings are $8.65(3)$ and $3.789(2)$ Å, respectively [symmetry code: (i) $-x, 1-y, 1-z$]. The distances between the mean planes of the $\text{C1A}-\text{C6A}$ and $\text{C7A}^i-\text{C12A}^i$ rings, and $\text{C7A}-\text{C12A}$ and $\text{C1A}^i-\text{C6A}^i$ rings are $3.632(3)$ and $3.494(2)$ Å, respectively. For molecule *B*, the dihedral angle and the centroid-to-centroid distance between the $\text{C1B}-\text{C6B}$ and $\text{C7B}^{ii}-\text{C12B}^{ii}$ benzene rings are $3.64(2)$ and $3.737(2)$ Å, respectively [symmetry code: (ii) $2-x, 2-y,$

$-z$]. The distances between the mean planes of $\text{C1B}-\text{C6B}$ and $\text{C7B}^{ii}-\text{C12B}^{ii}$, and $\text{C7B}-\text{C12B}$ and $\text{C1B}^i-\text{C6B}^i$ rings are $3.469(2)$ and $3.374(2)$ Å, respectively. These findings suggest the existence of $\pi-\pi$ stacking between the aromatic rings (Janiak, 2000) in the crystal structure of (I).

Experimental

Commercially available reagents, *viz.* aniline and 2,4-dinitrobenzaldehyde, were used without further purification. Aniline (0.5 ml) was added to a solution of 2,4-dinitrobenzaldehyde (0.005 mol) in hot methanol (40 ml). The reaction mixture was refluxed for 4 h. The title compound spontaneously crystallized from the warm methanol. The crystals were filtered off, washed in methanol and dried in an air (m.p. 410–411 K). Calculated: C 57.56, H 3.32, N 15.49%; found: C 57.50, H 3.34, N 15.22%.

Crystal data

$\text{C}_{13}\text{H}_9\text{N}_3\text{O}_4$	$Z = 4$
$M_r = 271.23$	$D_x = 1.489 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 7.5001(8) \text{ \AA}$	Cell parameters from 3726 reflections
$b = 11.9920(13) \text{ \AA}$	$\theta = 3.1-36.6^\circ$
$c = 14.1110(14) \text{ \AA}$	$\mu = 0.11 \text{ mm}^{-1}$
$\alpha = 90.867(10)^\circ$	$T = 100(2) \text{ K}$
$\beta = 93.531(7)^\circ$	Plate, brown
$\gamma = 107.084(8)^\circ$	$0.3 \times 0.2 \times 0.1 \text{ mm}$
$V = 1210.1(2) \text{ \AA}^3$	

Data collection

Kuma KM-4-CCD κ -geometry diffractometer	3926 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.052$
Absorption correction: none	$\theta_{\text{max}} = 28.5^\circ$
14484 measured reflections	$h = -9 \rightarrow 9$
6029 independent reflections	$k = -16 \rightarrow 15$
	$l = -16 \rightarrow 18$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.043$	$w = 1/[\sigma^2(F_o^2) + (0.03P)^2]$
$wR(F^2) = 0.086$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.05$	$(\Delta/\sigma)_{\text{max}} = 0.045$
6029 reflections	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
433 parameters	$\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

$\text{C7A}-\text{C14A}$	1.490(2)	$\text{C7B}-\text{C14B}$	1.488(2)
$\text{N14A}-\text{C14A}$	1.2690(19)	$\text{N14B}-\text{C14B}$	1.271(2)
$\text{N14A}-\text{C6A}$	1.4260(19)	$\text{N14B}-\text{C6B}$	1.4211(19)
$\text{C7A}-\text{C8A}-\text{N1A}-\text{O2A}$	$-19.7(2)$	$\text{C7B}-\text{C8B}-\text{N1B}-\text{O2B}$	$-16.8(2)$
$\text{C6A}-\text{N14A}-\text{C14A}-\text{C7A}$	$175.34(12)$	$\text{C6B}-\text{N14B}-\text{C14B}-\text{C7B}$	$176.98(13)$

After their location in a difference Fourier map, H atoms were placed in calculated positions and allowed to ride on their parent atoms, with $\text{C}-\text{H} = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2001); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2001); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXL97*.

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